

Admission test -Degree Course in Pharmacy, University of Rome 'Tor Vergata', Academic Year 2017/2018

- 1** When the anion MnO_4^- is transformed into cation Mn^{2+} , manganese:
- A Goes from oxidation number +7 to +2 and is reduced
B Goes from oxidation number +7 to +2 and is oxidized
C Goes from oxidation number -1 to +2 and is reduced
D Goes from oxidation number -2 to +2 and is reduced
E None of the above
- 2** Indicate which is the sulfite ion:
- A SO_4^{2-}
B SO_3^{2-}
C S^{2+}
D $\text{S}_2\text{O}_3^{2-}$
E S^{2-}
- 3** Given that the atomic mass of carbon is 12, how many atoms are present in 0,36g of carbon?
- A 50
B 0,03
C $6,022 \times 10^{23}$
D $1,8066 \times 10^{22}$
E 12
- 4** If in a water solution the concentration of H_3O^+ ions is 0,001M, what is the pH?
- A 1
B 0,001
C 3
D 30
E 7
- 5** How many grams of ethyl alcohol ($\text{C}_2\text{H}_5\text{OH}$, MW 46,1) are present in 500mL of a solution 0,2M?
- A 4,61g
B $6,022 \times 10^{23}\text{g}$
C $1,0 \times 10^{23}\text{g}$
D 0,461g
E None of the above
- 6** Methanol is:
- A An organic compound containing a hydroxy group
B An aromatic organic compound
C A halogenated organic compound
D An organic compound containing an amino group
E None of the above
- 7** An atom that contains 19 protons, 20 neutrons and 19 electrons has the following mass number:
- A 19
B 20
C 39
D 58
E 19,5
- 8** Which of the following solutions has $\text{pH} > 7$?
- A 50mL of a 0,1M solution of NaCl
B 50mL of a 0,1M solution of HCl
C 250mL of a 0,1M solution of HCl
D 50mL of a 0,1M solution of NaOH
E 50mL of a 0,1M solution of NH_4Cl
- 9** A 0,5 M solution of KCl contains:
- A 1mol of solute per liter of solution
B 0,5g of solute per liter of solution
C 0,5g of solute per Kg of solution
D 1mole of solute per Kg of solution
E 0,5moles of solute per liter of solution
- 10** Which of the following couples of atoms can form a covalent bonding?
- A H; Na
B H; Br
C Na; Br
D Ca; Cl
E Ca; O

- 11 A gas sample at $T=0^{\circ}\text{C}$ and $P= 1 \text{ atm}$ has volume $V= 44,8 \text{ L}$, contains:**
- A About 1 mole
 - B About 1gram
 - C About 2 moles
 - D About 2grams
 - E About 2 molecules
- 12 Ice and water constitute a system:**
- A Chemically heterogeneous and physically heterogeneous
 - B Chemically homogeneous and physically heterogeneous
 - C Chemically homogeneous and physically homogeneous
 - D Chemically heterogeneous and physically homogeneous
 - E None of the above
- 13 Which species can be formed upon combustion of methane (CH_4)?**
- A Hydrogen and oxygen
 - B Carbon and hydrogen
 - C Propane
 - D Methane gas
 - E Carbon dioxide and water
- 14 In order to neutralize 100mL of a 0,01M solution of KOH it is necessary to add:**
- A 0,1mL of H_2O
 - B 0,1mL of a 1 M solution of NaOH
 - C 1mL of a 10^{-2} M solution of HNO_3
 - D 100mL of a 10^{-2} M solution of HNO_3
 - E 100mL of a 10^{-2} M solution of NaOH
- 15 In order to neutralize 100ml of a 10^{-3}M solution of HCl it is necessary to add:**
- A 1000ml of H_2O
 - B 10ml of a 0.01 M solution of NaOH
 - C 10ml of a 0.01 M solution of KCl
 - D 1ml of a 10^{-2} M solution of HNO_3
 - E 1ml of a 10^{-2} M solution of NaOH
- 16 In eukaryotic cells, DNA occurs only in:**
- A Cell nuclei, mitochondria, Golgi apparatus
 - B Mitochondria, Golgi apparatus, chloroplasts
 - C Cell nuclei, lysosomes, chloroplasts
 - D Cell nuclei, mitochondria, chloroplasts
 - E Mitochondria, chloroplasts, lysosomes
- 17 Two parents who do not exhibit phenylketonuria (PKU) have a son with PKU. Which of the following conclusions can be drawn from this situation?**
- A The allele for PKU is located on the Y chromosome
 - B PKU is a dominant trait
 - C PKU is a recessive trait
 - D A mutation occurred in the sperm of the father
 - E None of the above
- 18 Which of the following cellular processes normally produces ATP from glucose in the absence of oxygen?**
- A Krebs cycle
 - B Glycolysis
 - C Chemiosmosis
 - D Calvin cycle
 - E None of the above
- 19 Which of the following best describes the pathway of a protein from its manufacture to its secretion from the cell?**
- A Endoplasmic reticulum \rightarrow Golgi complex \rightarrow secretory vesicle
 - B Secretory vesicle \rightarrow endoplasmic reticulum \rightarrow Golgi complex
 - C Secretory vesicle \rightarrow Golgi complex \rightarrow endoplasmic reticulum
 - D Golgi complex \rightarrow endoplasmic reticulum \rightarrow secretory vesicle
 - E None of the above
- 20 A person touches a hot object and immediately moves her finger away from it. Which of the following structures is the first to receive an impulse triggered by the stimulus?**
- A Synapse
 - B Ventral root ganglion
 - C Motor neuron
 - D Sensory neuron
 - E None of the above

- 21 The sympathetic nervous system**
 A Increases heart rate
 B Is voluntary
 C Is activated mainly in sleep
 D Activates digestion
 E None of the above
- 22 A health disorder that results from a deficiency of thyroid hormone production in adults and is characterized by a low metabolic rate, increase in body weight and tiredness is:**
 A Hyperthyroidism
 B Goiter
 C Cretinism
 D Hypothyroidism
 E None of the above
- 23 In eukaryotic cells, protein synthesis takes place at the:**
 A Nucleus
 B Mitochondria
 C Chloroplasts
 D Ribosomes
 E Lysosome
- 24 The four bases which form the code words for DNA are:**
 A UTAC
 B ACTU
 C AGCU
 D ATCG
 E None of the above
- 25 Prokaryotic cells are different from eukaryotic cells because prokaryotic cells:**
 A Are much bigger
 B Have no cell nuclei
 C Have cell nuclei
 D Have no cell wall
 E Have mitochondria and chloroplasts
- 26 The DNA double helix is stabilized by hydrogen bonds between:**
 A Purine bases and complementary pyrimidine bases
 B Pyrimidine bases only
 C Purine bases only
 D Molecules of deoxyribose and phosphate groups
 E Nitrogenous bases and phosphate groups
- 27 During which of the following stages of cell cycle DNA replication occurs?**
 A Early prophase
 B G1 phase
 C G2 phase
 D G0 phase
 E S phase
- 28 Which of the following statements best describes mitosis?**
 A It occurs only in the reproductive structure of organism
 B During this stage DNA is synthesized
 C It is one of the interphase stages
 D It is the division of mother cell into two daughter cells, genetically identical to each other and to their parent cell
 E It is the division of mother cell into two daughter cells, genetically different from each other and from their parent cell
- 29 Gene expression may be regulated:**
 A Only during translation
 B Only during transcription
 C During transcription and translation
 D Only during replication
 E During replication and transcription
- 30 One of a series of different versions of gene is called:**
 A Genome
 B Haploid
 C Diploid
 D Allele
 E Allosome

- 31 Which of the following processes can decrease the genetic variation in natural populations?**
 A Recombination
 B Immigration
 C Mutagenesis
 D Inbreeding
 E Hybridization
- 32 Eutrophication is the process of:**
 A Vegetation succession in lakes
 B Nutrient depletion of lakes
 C Increase inflow of minerals in lakes
 D Primary succession in lakes
 E Biological renewal of aquatic ecosystems
- 33 Which of the following types of plants tissue transport water and minerals?**
 A Phloem
 B Xylem
 C Meristem
 D Vascular cambium
 E Epidermis
- 34 Which of the following require a host cell because they are notable to make proteins on their own?**
 A Paramecium caudatum
 B Escherichia coli
 C Influenza virus
 D Saccharomyces cerevisiae
 E Euglena viridis
- 35 Which of the following is found in an unusual high proportion in cells of cardiac muscle?**
 A Lysosomes
 B Mitochondria
 C mRNA
 D Golgi apparatus
 E Cell nucleus
- 36 Which of the following organisms conduct ethanol fermentation?**
 A Yeasts
 B Bears
 C Spiders
 D Barley
 E Mushrooms
- 37 Which of the following is the right order of the four stages of food processing?**
 A Digestion, ingestion, absorption, elimination
 B Ingestion, digestion, elimination, absorption
 C Ingestion, digestion, absorption, elimination
 D Digestion, ingestion, absorption, elimination
 E Absorption, elimination, digestion, ingestion
- 38 Which hormone is produced by the placenta?**
 A Progesterone
 B Insulin
 C Thyroxin
 D Melatonin
 E None of the above
- 39 Which of the following is not characteristic for mammals?**
 A Mammary glands
 B Hair
 C Cold-blooded
 D Three middle ears bones
 E Sweat glands
- 40 All of the members of a particular species that live in one area are called a(an):**
 A Biotope
 B Community
 C Ecosystem
 D Biosphere
 E Population

- 41 Consider the following inequalities:**
- A $(2/3) > (3/4)$
 B $(2/3) < (2/5)$
 C $z^{-1} > z^{-2}$ when z is real and $z > 1$
 D $2 + \frac{1}{10} > 1 + \frac{11}{10}$
 E None of the above
- 42 What is the solution of the linear system $x+y=5$; $2x+3y=4$?**
- A $x=1, y=1$
 B $x=11, y=-6$
 C The linear system has infinite solutions
 D $x=-11, y=6$
 E The linear system has no solution
- 43 What is/are the correct value(s) of $(a + b)^2$ (a, b real numbers)?**
- A $a^2 + ab + b^2$
 B $a^2 + b^2$
 C $a^2 - b^2$
 D $a^2 + 2ab + b^2$
 E $a^2 + 2ab + b^2$
- 44 Consider the equation $4 - \sqrt{7 + \sqrt{9 + \sqrt{4 + x}}} = 0$. Mark the correct value of x**
- A 3660
 B 46
 C 567
 D 68
 E 5180
- 45 What is/are the correct value(s) of x for the equation $\log_x(16) = \frac{4}{3}$?**
- A $x = \sqrt[3]{65536}$
 B $x = 64$
 C $x = 2$
 D $x = 8$
 E $x = 1$
- 46 Calculates the mass of the motorcycle with speeds of 10 m / s and kinetic energy of 10000 J**
- A 100 kg
 B 300 kg
 C 200 kg
 D 400 kg
 E 500 kg
- 47 Which physical quantity can be measured in Newton meter (N m)?**
- A Force
 B Momentum
 C Kinetic energy
 D Power
 E Work
- 48 Which physical quantity can be measured in Newton second (N s)?**
- A Force
 B Momentum
 C Kinetic energy
 D Power
 E Work
- 49 When do you have a uniform circular motion?**
- A When there is a variable centripetal force
 B When there is a constant centripetal force
 C When there is a constant centrifugal force
 D When there is a variable centrifugal force
 E When there is a constant velocity
- 50 A net force F_A acts on object A, and a net force F_B acts on object B. The mass of object B is twice the mass of object A, and the acceleration of object B is twice that of object A. Which of the following relations between F_A and F_B is true?**
- A $F_B = \frac{1}{4} F_A$
 B $F_B = \frac{1}{2} F_A$
 C $F_B = F_A$
 D $F_B = 2 F_A$
 E $F_B = 4 F_A$

- 51 In which year the Wright brothers achieved the first powered, sustained and controlled airplane flight?**
 A 1908
 B 1918
 C 1880
 D 1903
 E None of the above
- 52 The Machu Picchu is an archeological site located in?**
 A Perú
 B Iran
 C Mexico
 D Colombia
 E None of the above
- 53 Which is the Capital of Romania?**
 A Budapest
 B Bucarest
 C Belgrad
 D Novi Sad
 E None of the above
- 54 Which architect designed the Sagrada Familia in Barcellona?**
 A Le Corbusier
 B Carlos Gardel
 C Antoni' Gaudi'
 D Oscar Niemeyer
 E None of the above
- 55 Who first discovered the oral polio vaccine?**
 A Fleming
 B Sabin
 C Dulbecco
 D Pasteur
 E None of the above
- 56 In which year Christopher Columbus discovered America?**
 A 1592
 B 1472
 C 1492
 E 1542
- 57 Who designed the cupola of Santa Maria del Fiore in Florence?**
 A Michelangelo
 B Raffaello
 C Maderno
 D Brunelleschi
 E None of the above
- 58 Which of the following movies was not directed by Stanley Kubrick?**
 A 2001: A Space Odyssey
 B Lawrence of Arabia
 C Shining
 D A clockwork Orange
 E None of the above
- 59 Which of the following Nations is not a Republic?**
 A Portugal
 B Austria
 C Greece
 D Sweden
 E None of the above
- 60 Roger Federer is:**
 A Football player
 B Basketball player
 C Volleyball player
 D Tennis player
 E None of the above